

Synthesis and study of biological activities of compounds derived from new Imidazole derivative

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Abstract

Background: Heterocyclic compounds are of great importance for the life, because many natural products such as hormones and antibiotics have structural subunits. Imidazole derivatives were used as a precursor for synthesis of many heterocyclic systems. The aim of current study was to use the easily available benzoyl thiourea for the synthesis of Imidazole derivative Ethyl 2-((1-benzoyl-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate. Methods: the imidazole compounds of interest were synthesized using a range of precursors through addition, reflux and recrystallization reactions. Results: data obtained from current study revealed that the synthesized imidazole derivatives were characterised and their antibacterial activities were tested. These compounds exhibited powerful inhibition of growth against *P. aerogenosa* and *S. aureus* bacterial strains. Conclusions: Imidazole derivatives can be manufactured from available and cost-effective precursors to be employed as potent antibacterial agents.

Keyword: Benzoyl chloride, Thiourea, Imidazole, Chalcone, Schiff's base.

1. INTRODUCTION

The heterocyclic rings are of biological interest because of their potential chemical and physical properties [1]. heterocyclic compounds are of great importance to the life, because many natural products such as hormones and antibiotics have structural subunits [2,3]. The practical method of composition of these compounds used for synthesis of many organic compounds [4]. Nitrogen-containing heterogeneous compounds play an important role in medical chemistry by assisting in various biological processes [5]. In recent years imidazole derivatives containing nitrogen have attracted increased attention due to the broad spectrum of their biological activities [6]. Chalcone are still promising to conduct new drug analyzes. For this, new ways of synthesizing the alkaloid derivatives which exhibited a range of pharmacological and biological effects [7]. Chalcone derivatives were reported to have a broad spectra of biological activities e.g. antimalarial [8,9], anticancer [10,11], antioxidant, anti-inflammatory [12-15], antimicrobial [16,17], antifilarial [18], antifungal [19, 20], larvicidal [21], and anticonvulsant activities [22]. Schiff's bases were derived from some heterocyclic compounds which have broad spectra of biological activities e.g. antiviral, anti-glycation, anticonvulsant anti-inflammatory, antimicrobial, antidepressant, angiotensin-II receptor antagonist, and anticancer activity [23,24].

2. MATERIAL AND METHODS

2.1. Synthesis of 1-benzoyl-thiourea-[BT]

A solution of 2.28 g (0.03 mole) of thiourea in 25 ml 1,4-dioxane, was added drop-wise at room temperature to 1.40 g (0.01 mole) of benzoyl chloride, refluxed for 12 h, poured into ice-cold water, precipitated, filtered off and then solid recrystallized of suitable solvent to yield the wanted compound. Crude product was recrystallized from benzene.

(BT): White powder; yield 92%; m.p. 171-173 °C.

IR (ν/cm^{-1}): ν_{NH_2} = 3308, ν_{NH} = 3232, $\nu_{\text{C=O}}$ amide = 1676, $\nu_{\text{C=S}}$ = 1103.

¹H-NMR: δ (DMSO-d₆): 11.27(s,SH,1H), 9.88 (s,NH=CS,1H), 9.60 (s, CONH, 1H), 7.39- 8.01 (m, 5H, Aromatic).

MS, m/z [M]⁺: 180 (180.3) found (calcu.).

2.2. Synthesis of ethyl 2-((1-benzoyl-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate (1):

Benzoyl thioureas 1.8 g (0.01 mole) was dissolved in 30 ml of 1,4-dioxane. To this solution, anhydrous potassium carbonates 1.37g (0.01 mole), was added as a catalyst, after that ethyl chloroacetate (2.25 ml; 0.02mole) was added to above solution and stirred well. The mixture was refluxed for 7 hr. After that this

solution was added to crush ice to get the solid. TLC indicates the presence of some impurities and recrystallized from water: ethanol (7:3) ml.

(1):- Red powder; yield=75%; m.p. 123-125 °C.

IR (ν/cm^{-1}): 3026 (C-Har), 2856-2933 (aliphatic C-H), $\nu_{\text{C=O}}$ ester = 1773, $\nu_{\text{C=O}}$ keton = 1687, $\nu_{\text{C=O}}$ amide = 1639, $\nu_{\text{C=N}}$ = 1599.

¹H-RMN: δ H (DMSO-d₆): 1.2 (t, CH₃-CH₂,3H), 4.6 (s, S-CH₂,2H), 4.4 (q, COO-CH₂,2H), 4.4 (s, CO-CH₂,2H), 7.2- 8.1 (m, 5H, Aromatic).

MS, m/z [M]⁺: 306 (306.07) found (calcu.)

2.3. Synthesis of ethyl 2-((1-benzoyl (Arylidene)-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate [2a-e]

To a solution of compound (1), 1g (0.0032 moles) in 30 ml ethanol aromatic aldehydes was added (o-bromobenzaldehyde, p-nitrobenzaldehyde, 4-dimethylamino benzaldehyde, benzaldehyde and anisaldehyde) (0.0032 mole) and 4 ml of 10% KOH solution and stirred for 24 hr at 25 °C. Then, ice-water was poured into precipitate with few drops of hydrochloric acid. The product was recrystallized from suitable solvent afford to get the compound of interest.

Ethyl2-((1-benzoyl-4-(2-bromobenzylidene)-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate(2a): Black powder; yield=68%; m.p. 150-153 °C.

IR (ν/cm^{-1}): 3068 (C-Har), 2872-2933 (C-H aliph), $\nu_{\text{C=O}}$ ester = 1739, $\nu_{\text{C=O}}$ keton = 1718, $\nu_{\text{C=O}}$ amide = 1647, $\nu_{\text{C=N}}$ = 1599, $\nu_{\text{C=CH}}$ = 1579, $\nu_{\text{C-Br}}$ = 759.

¹H-RMN: δ H (DMSO-d₆): 1.2 (t, CH₃-CH₂,3H), 4.2 (q, COO-CH₂,2H), 4.8 (s, S-CH₂,2H), 7.4- 8.0 (m, 9H, Aromatic), 8.2 (s, C=CH,1H).

MS, m/z [M]⁺: 474 (473.34) found (calcu.).

Ethyl2-((1-benzoyl-4-(4-nitrobenzylidene)-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate(2b): Yellow powder; yield=70%; m.p. =192-194 °C.

IR (ν/cm^{-1}): 3086 (C-H ar), 2921-2983 (C-H aliph), $\nu_{\text{C=O}}$ ester = 1745, $\nu_{\text{C=O}}$ keton = 1716, $\nu_{\text{C=O}}$ amide = 1645, $\nu_{\text{C=N}}$ = 1597, $\nu_{\text{C=CH}}$ = 1577, ν_{NO_2} = 1533-1342.

¹H-RMN: δ H (DMSO-d₆): 1.2 (t, CH₃-CH₂,3H), 4.4 (q, COO-CH₂,2H), 5.0 (s, S-CH₂,2H), 7.4- 8.0 (m, 9H, Aromatic), 8.6 (s, C=CH,1H).

MS, m/z [M]⁺: 439 (439.08) found (calcu.).

Ethyl-((1-benzoyl-4-(4-(dimethylamino)benzylidene)-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate(2c): Orang powder; yield=78%; m.p.160-163 C°.

IR (ν/cm^{-1}): 3063 (C-Har), 2810-2908 (C-H aliph), $\nu_{\text{C=O}}$ ester = 1743, $\nu_{\text{C=O}}$ keton = 1705, $\nu_{\text{C=O}}$ amide = 1637, $\nu_{\text{C=N}}$ =1616, $\nu_{\text{C=CH}}$ =1527, $\nu_{\text{p-N(CH}_3)_2}$ = 804.

¹H-RMN: δ H (DMSO-d6): 1.3 (t, CH₃-CH₂,3H), 4.2 (q, COO-CH₂,2H), 4.9 (s, S-CH₂,2H), 3.1 (s, N(CH₃)₂,6H), 7.62- 7.9 (m, 9H, Aromatic), 8.2 (s, C=CH,1H).

MS, m/z [M]⁺: 437 (437.14) found (calcu.).

Ethyl-((1-benzoyl-4-benzylidene-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio) acetate(2d): White powder; yied=70%; m.p. =164-167 C°.

IR (ν/cm^{-1}): 3066 (aromatic C-H), 2872-2983 (aliphatic C-H), $\nu_{\text{C=O}}$ ester = 1737, $\nu_{\text{C=O}}$ keton = 1707, $\nu_{\text{C=O}}$ amide = 1647, $\nu_{\text{C=N}}$ =1599 $\nu_{\text{C=CH}}$ =1577.

¹H-RMN: δ H (DMSO-d6): 1.4 (t, CH₃-CH₂,3H), 4.1 (q, COO-CH₂,2H), 4.8 (s, S-CH₂,2H), 7.5- 7.7 (m, 10H, Aromatic), 8.2 (s, C=CH,1H).

MS, m/z [M]⁺: 393 (393.10) found (calcu.).

Ethyl-((1-benzoyl-4-(4-methoxybenzylidene)-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetate(2e): Yellow powder; yield=55%; m.p. 140-145 C°.

IR (ν/cm^{-1}): 3064 (C-Har), 2841-2983 (C-H aliph), $\nu_{\text{C=O}}$ ester = 1739, $\nu_{\text{C=O}}$ keton = 1712, $\nu_{\text{C=O}}$ amide = 1643, $\nu_{\text{C=N}}$ =1593, $\nu_{\text{C=CH}}$ =1575, $\nu_{\text{p-OH}_3}$ = 827.

¹H-RMN: δ H (DMSO-d6): 1.2 (t, CH₃-CH₂,3H), 4.2 (q, COO-CH₂,2H), 4.8 (s, S-CH₂,2H), 3.8 (s, OCH₃,3H), 7.1- 7.9 (m, 9H, Aromatic), 8.2 (s, C=CH,1H).

2.4. Synthesis of 2-((1-benzoyl-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetohydrazide-3

To solution (1), 1g (0.0032 moles) in ethanol, hydrazine hydrate 80%, (0.32 g; 0.0065 moles) were added. Then refluxed for 10 hr and the obtained product was crystallized from ethanol.

2-((1-benzoyl-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio)acetohydrazide(3):

White powder; yied=60%; m.p. =245-250 C°.

IR (ν/cm^{-1}): 3039 (C-H ar), 2839-2953 (C-Haliph), ν_{NH_2} = 3298, ν_{NH} = 3186, $\nu_{\text{C=O}}$ keton = 1662, $\nu_{\text{C=O}}$ amide = 1639, $\nu_{\text{C=N}}$ =1590.

¹H-RMN: δ H (DMSO-d6): 3.8 (s, CO-CH₂,2H), 4.2 (s, S-CH₂,2H), 5.3 (s, NH₂,2H), 7.1- 7.9 (m, 5H, Aromatic), 12.1 (s, NH,1H).

MS, m/z [M]⁺: 294 (292.06) found (calcu.).

2.5. Synthesis of 2-((1-benzoyl-5-oxo-4,5-dihydro-1H-imidazol-2-yl)thio) (Arylidene)acetohydrazide- (4a-e)

To a solution of acid hyrazide (3), 2.92 g (0.01 mol) in methanol (30 ml with few drops of acetic acid, 0.01 mol of aromatic aldehydes (o-bromobenzaldehyde, p-nitrobenzaldehyde, 4-dimethylaminobenzaldehyde, benzaldehyde and anisaldehyde) were added, refluxed for 6 hr, then cooled and filtered. The obtained product was recrystallized from ethanol.

2-((1-benzoyl-5oxo-4,5-dihydro1H-imidazol-2-yl)thio)-N'-(2-bromobenzylidene) acetohydrazide(4a): white powder; yield=70%; m.p.187-190 C°.

IR (ν/cm^{-1}): 3082 (C-Har), 2854-2962 (C-H aliph), ν_{NH} = 3207, $\nu_{\text{C=O}}$ keton = 1683, $\nu_{\text{C=O}}$ amide = 1645, $\nu_{\text{C=N}}$ =1599, $\nu_{\text{N=CH}}$ = 1620.

¹H-RMN: δ H (DMSO-d6): 4.6 (s, COCH₂,2H), 4.6 (s, S-CH₂,2H), 7.2-7.9 (m, 9H, Aromatic), 8.5 (s, N=CH,1H), 12.1 (s, NH,1H).

2-((1-benzoyl-5oxo-4,5-dihydro1H-imidazol-2-yl)thio)-N'-(4-nitrobenzylidene) acetohydrazide(4b): yellow powder; yield=75%; m.p. 278-280 C°.

IR (ν/cm^{-1}): 3061 (aromatic C-H), 2939 (aliphatic C-H), ν_{NH} = 3207, $\nu_{\text{C=O}}$ keton = 1678, $\nu_{\text{C=O}}$ amide = 1639, $\nu_{\text{C=N}}$ =1599, $\nu_{\text{N=CH}}$ = 1629, ν_{NO_2} =1521-1344.

¹H-RMN: δ H (DMSO-d6): 3.9 (s, COCH₂,2H), 4.3 (s, S-CH₂,2H), 7.5-7.9 (m, 9H, Aromatic), 8.8 (s, N=CH,1H), 12.1 (s, NH,1H).

MS, m/z [M]⁺: 426 (425.08) found (calcu.).

2-((1-benzoyl-5oxo-4,5dihydro1H-imidazol-2-yl)thio)-N'-(4-(dimethylamino) benzylidene)acetohydrazide(4c): Orang powder; yied=65%; m.p. =268-270 C°.

IR (ν/cm^{-1}): 3068 (C-Har), 2866-3020 (C-H aliph), ν_{NH} = 3213, $\nu_{\text{C=O}}$ keton = 1674, $\nu_{\text{C=O}}$ amide = 1641, $\nu_{\text{C=N}}$ =1599, $\nu_{\text{N=CH}}$ = 1641, $\nu_{\text{p-N(CH}_3)_2}$ = 812.

¹H-RMN: δ H (DMSO-d6): 2.9 (s, N(CH₃)₂,6H), 3.9 (s, COCH₂,2H), 4.3 (s, S-CH₂,2H), 7.3-7.9 (m, 9H, Aromatic), 8.1 (s, N=CH,1H), 11.2 (s, NH,1H), 12.2 (s, OH,1H).

2-((1-benzoyl-5oxo-4,5dihydro1H-imidazol-2-yl)thio)-N'-benzylidene acetohydrazide(4d): White powder; yield=60%; m.p. 200-203 C°.

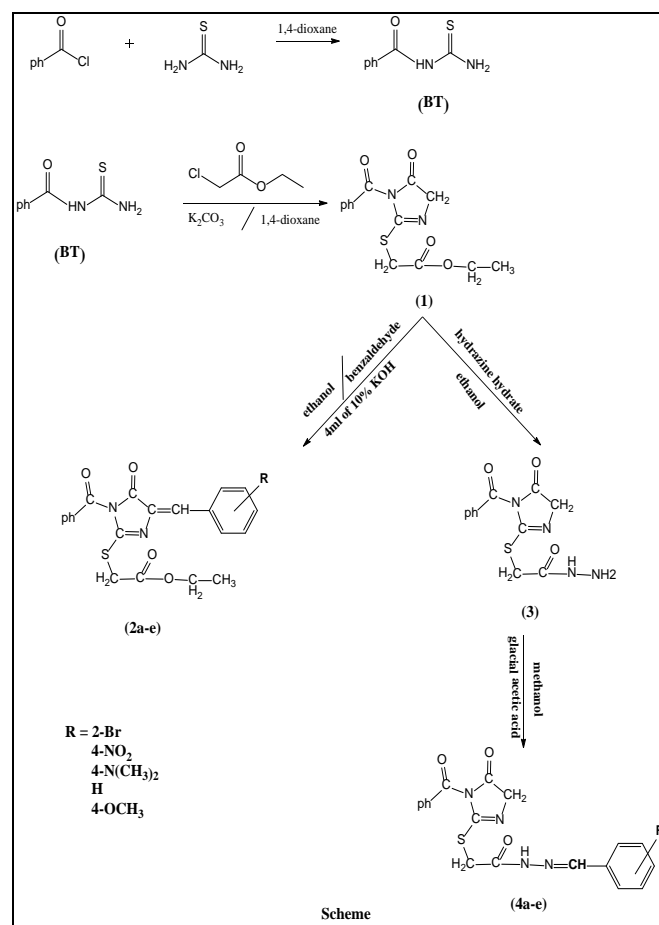
IR (ν/cm^{-1}): 3078 (C-Har), 2810-3016 (C-Haliph), ν_{NH} = 3213, $\nu_{\text{C=O}}$ keton = 1680, $\nu_{\text{C=O}}$ amide = 1641, $\nu_{\text{C=N}}$ =1606, $\nu_{\text{N=CH}}$ = 1622.

¹H-RMN: δ H (DMSO-d6): 4.50(s, COCH₂,2H), 4.50 (s, S-CH₂,2H), 7.30-7.90 (m, 10H, Aromatic), 8.22 (s, N=CH,1H), 11.41 (s, NH,1H).

2-((1-benzoyl-5oxo-4,5dihydro1H-imidazol-2-yl)thio)-N'-(4-methoxy benzylidene)acetohydrazide(4e): white powder; yied=65%; m.p. 185-190 C°.

IR (ν/cm^{-1}): 3088 (C-Har), 2847-3003 (C-Haliph), ν_{NH} = 3211, $\nu_{\text{C=O}}$ keton = 1678, $\nu_{\text{C=O}}$ amide = 1633, $\nu_{\text{C=N}}$ =1606, $\nu_{\text{N=CH}}$ = 1626, $\nu_{\text{p-OH}_3}$ = 829.

¹H-RMN: δ H (DMSO-d6): 3.82 (s, OCH₃,3H), 3.91 (s, COCH₂,2H), 4.5 (s, S-CH₂,2H), 7.0-7.9 (m, 9H, Aromatic), 8.1 (s, N=CH,1H), 11.3 (s, NH,1H).



3. RESULTS AND DISCUSSION

Our strategy was based on the use of easily available benzoyl thiourea which is synthesized from benzoyl chloride. The spectrum of IR of the benzoyl thiourea indicated a band at $(C=O)1676\text{ cm}^{-1}$ for carbonyl of amide, a band at 3308 cm^{-1} for (NH_2) and bands at 1103 cm^{-1} for $(C=S)$. The spectrum of ^1H-NMR for benzoyl thiourea showed a singlet at 11.27 ppm due to (SH) , singlet at 9.88 ppm due to $NH=CS$, singlet at 9.60 ppm due to $(CONH)$, and many signals at 7.39-8.01 ppm due to aromatic protons. The reaction of benzoyl thiourea with ethyl 2-chloroacetate and 1,4-dioxane in the presence of K_2CO_3 for the synthesis of Imidazole derivative [ethyl 2-((1-benzoyl-5oxo-4,5dihydro1H-imidazol-2yl)thio)acetate] (compound 1). IR spectrum of compound (1) showed disappearance of absorption bands for NH_2 , and a band of a $(C=O)$ of ester appeared at 1773 cm^{-1} for ester, a band at 1687 cm^{-1} for carbonyl of ketone and bands at 1639 cm^{-1} for carbonyl of amide. The ^1H-NMR spectrum for compound (1) showed signal as triplet at 1.21 ppm for (CH_3-CH_2) , singlet at 4.60 ppm due to $(S-CH_2)$, and a quartet at 4.41 ppm for $(COO-CH_2)$, singlet at 4.4 ppm due to $(CO-CH_2)$, and many signals at 7.2-8.1 ppm due to aromatic protons. Chalcone of compound (1) in ethanol was treated with some aldehydes with 10% KOH gave ethyl 2-((1-benzoyl (Arylidene)-5-oxo4,5-dihydro1H-imidazol-2yl)thio)acetate (2a-e). Compounds (2a-e) were confirmed with IR spectral and ^1H-NMR data. IR spectrum showed a band of $(C=CH)$ at $1575-1599\text{ cm}^{-1}$. While ^1H-NMR spectra for derivatives (2a-e) showed good signals, singlet at 8.2-8.6 ppm due to $(C=CH)$ and many signals at 7.1-8.0 ppm due to aromatic protons (5H s for phenyl ring, 4H dd for aryl ring). Ester, treated with hydrazine hydrate in ethanol, produced [2-((1-benzoyl-5oxo4,5-dihydro1H-imidazol-2yl)thio)acetohydrazide] (compound 3). The formation of compound (3) was confirmed by the disappearance of bands for ester and the presence of a band with 3298 cm^{-1} for (NH_2) and a band at 3186 cm^{-1} due to (NH) . While ^1H-NMR spectra showed singlet at 12.1 ppm due to (NH) and singlet at 5.3 ppm due to (NH_2) , compound (3) reacted with some aldehydes and yielded [2-((1-benzoyl-5oxo-4,5dihydro-1H-imidazol-2yl)thio)(Arylidene)aceto hydrazide] (4a-e). The Schiff's bases (4a-e) gave new IR absorption like $CH=N$ at $1621-1641\text{ cm}^{-1}$ and showed absence of (NH_2) stretching vibrations. ^1H-NMR spectra of (4a-e) showed singlet in 8.11-8.80 ppm due to $(N = CH)$ and many signals at 7.0-7.9 ppm due to aromatic protons (5H s for phenyl ring, 4H dd for aryl ring).

3.1. Antibacterial activity

The synthesized Imidazole derivatives were tested as antibacterial compounds using the agar disc-diffusion method against *Pseudomonas aeruginosa* and *Staph. aureus* bacteria. The control was DMSO which was used as a solvent and the derivatives concentration was $10^{-3}M$ (Table 1). Results of current study showed that all the tested derivatives had antibacterial activities against *Pseudomonas aeruginosa*. In addition, all tested derivatives, except compounds (3, 4a, 4b, 4c, and 4e), had antibacterial activities against *Staph. Aureus*. Moreover, compound (4d) showed high inhibition against *Pseudomonas aeruginosa* and *Staph. aureus* bacteria. In addition, compounds 2c, 3, 4a, 4c, 4d and 3e showed good inhibition against *Pseudomonas aeruginosa*.

Therefore, imidazole derivatives can be manufactured from available and cost-effective precursors to be employed as potent antibacterial agents.

Table 1 shows antibacterial activities of compounds 1-4e at a concentration of $10^{-3}M$

Comp. no.	<i>Pseudomonas aeruginosa</i> (Zone of bacterial growth inhibition/ mm)	<i>Staphylococcus aureus</i> (Zone of bacterial growth inhibition/ mm)
1	++	++
2 _a	++	++
2 _b	++	++
2 _c	+++	++
2 _d	++	++
2 _e	++	++
3	+++	-
4 _a	+++	-
4 _b	++	-
4 _c	+++	-
4 _d	+++	+++
4 _e	+++	-

- : no inhibition, +: 3-6 mm, ++: 7-10 mm and +++: 11-15 mm.

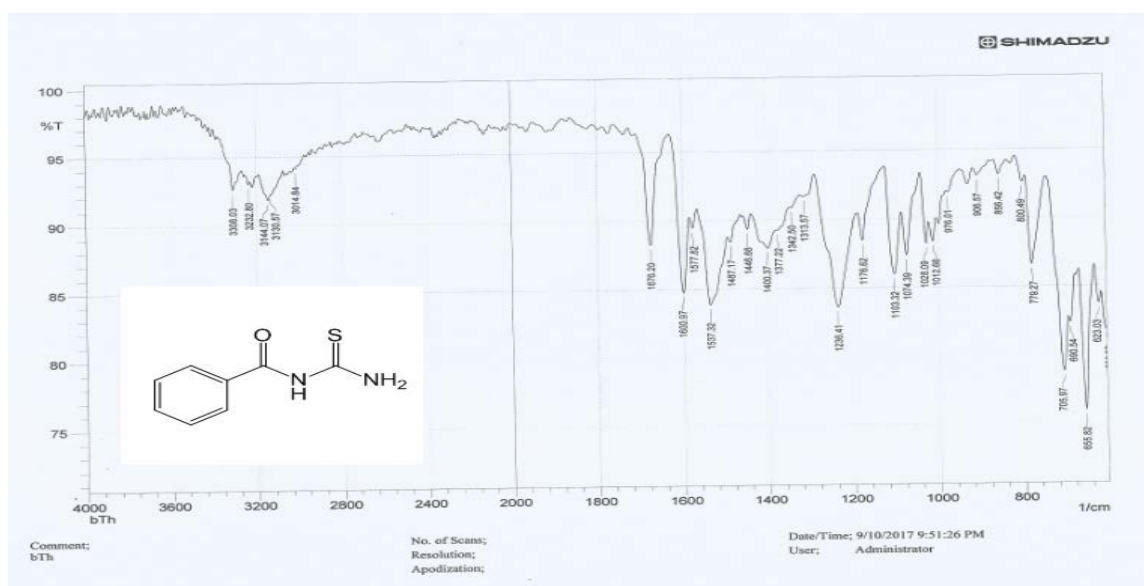


Figure 1 FTIR spectrum of (BT).

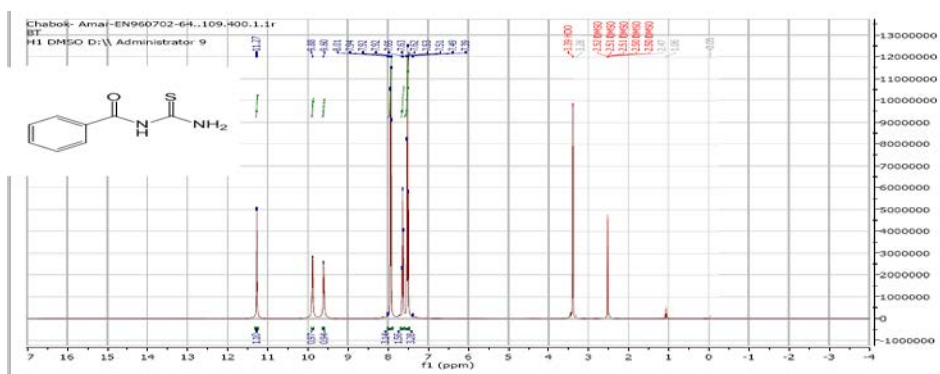


Figure 2 ¹H-NMR spectrum of (BT).

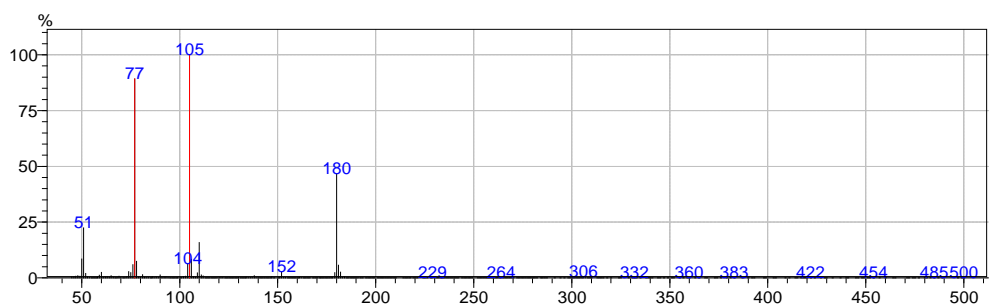


Figure 3 Mass spectrum of (BT).

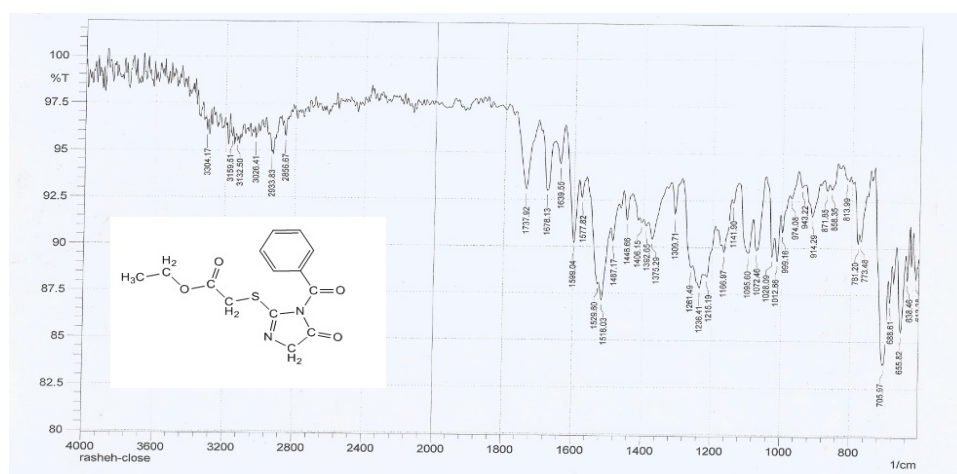


Figure 4 FTIR spectrum of compound 1.

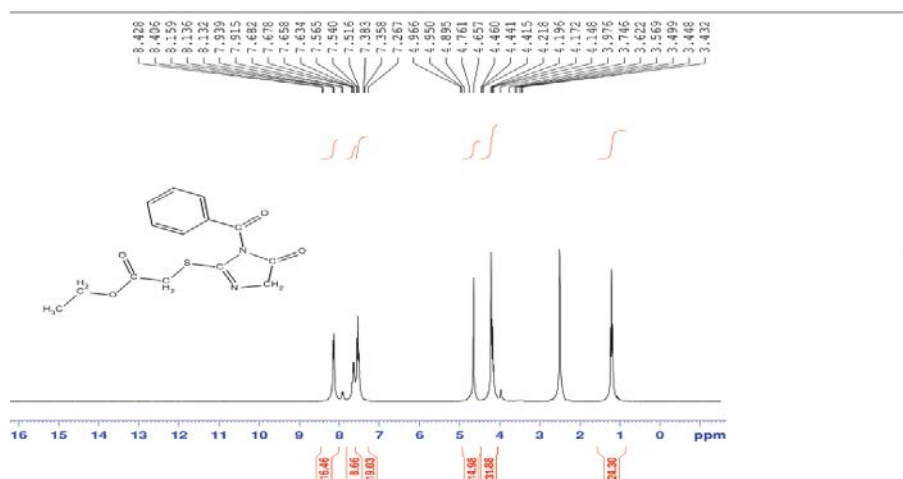


Figure 5 ¹H-NMR spectrum of compound 1.

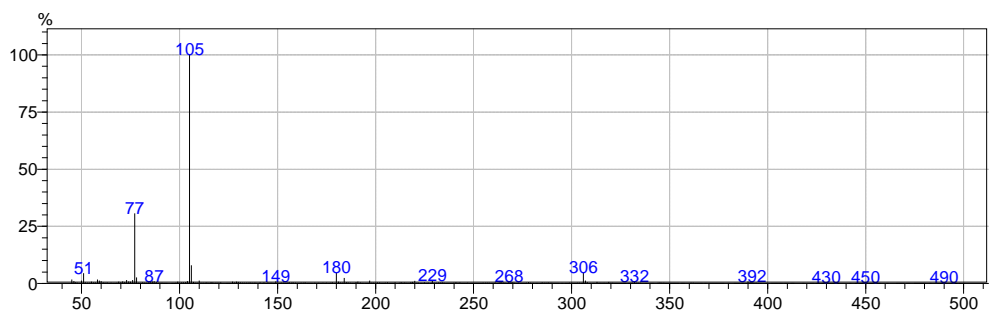


Figure 6 Mass spectrum of compound 1.

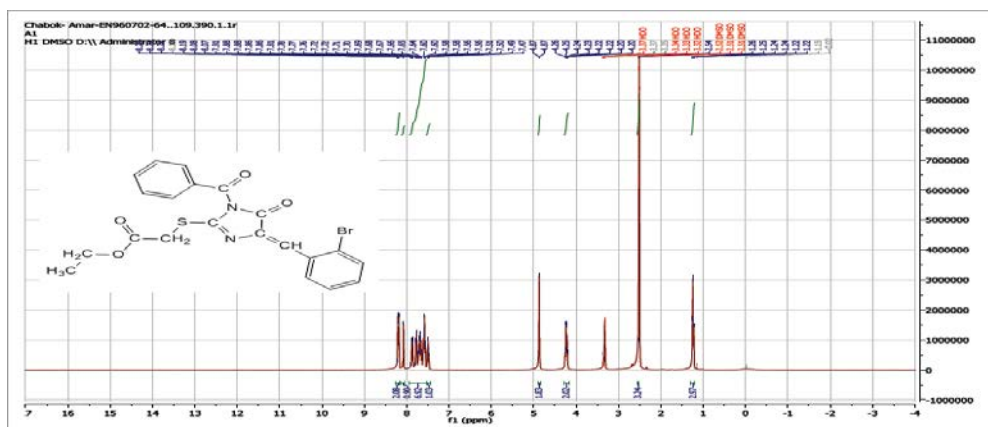


Figure 7 ¹H-NMR spectrum of compound 2a.

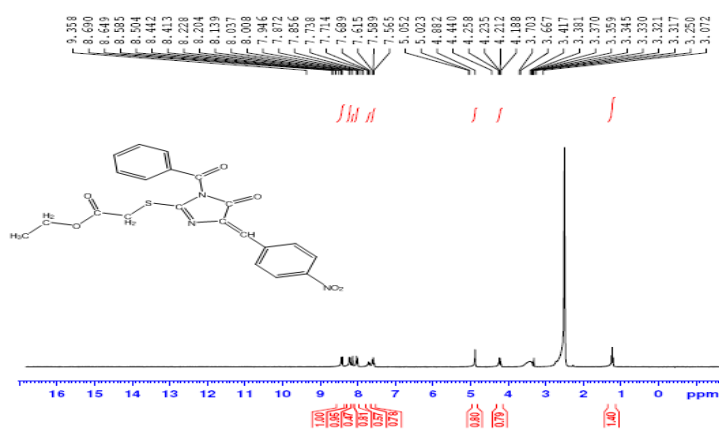


Figure 8 ¹H-NMR spectrum of compound 2b.

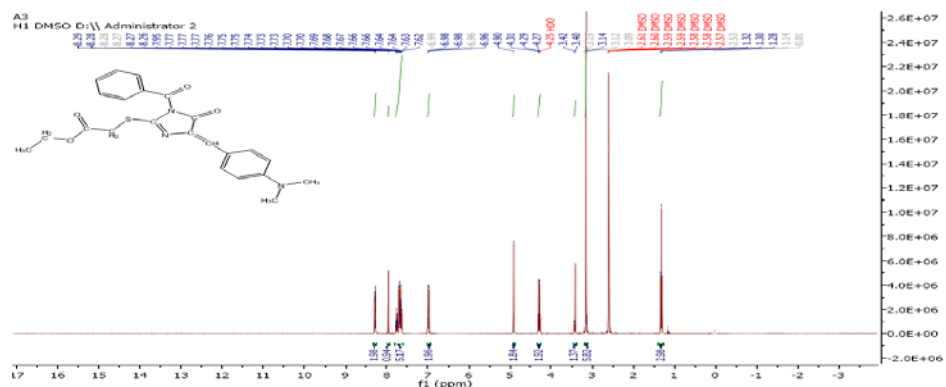
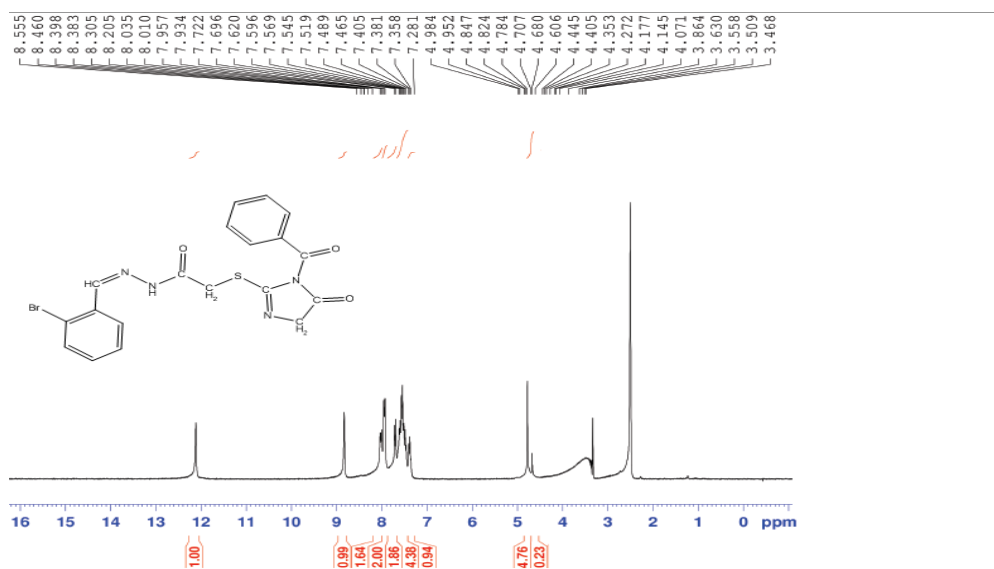
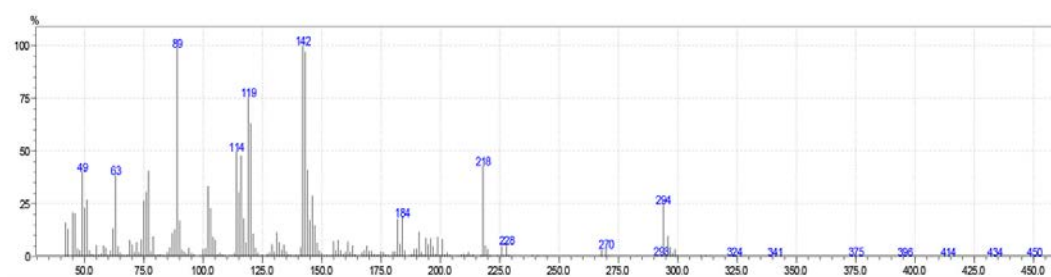
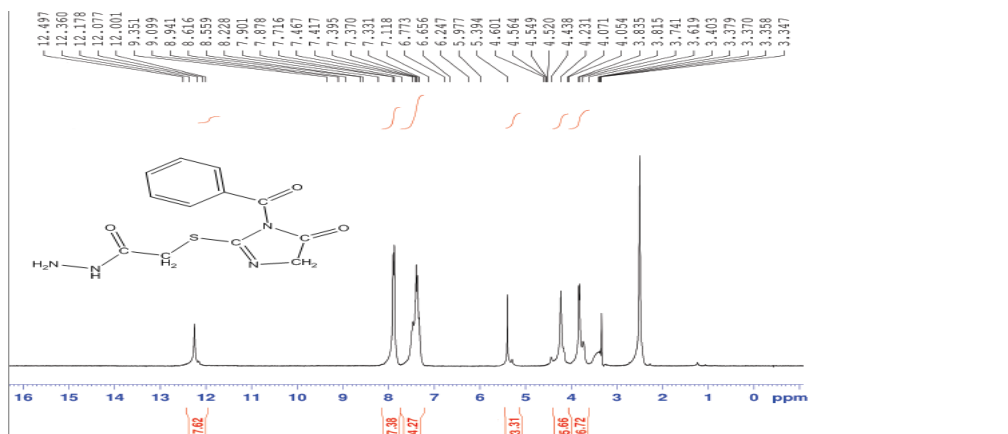
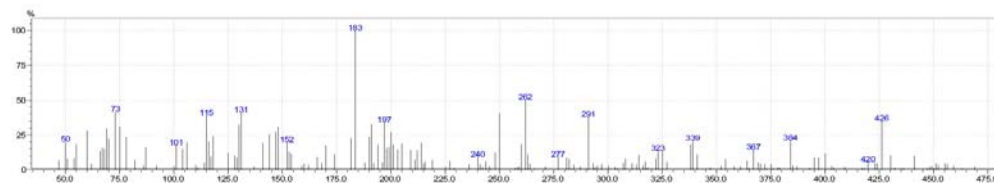


Figure 9 ¹H-NMR spectrum of compound 2c.



H2=425



REFERENCES

- Brown, R. C. D. Recent developments in solid-phase organic synthesis. *J. Chem. Soc. Perkin Trans.* 1998, 1, 3293–3320.
- Kachroo, M., Panda, R. & Yadav, Y. Synthesis and biological activities of some new pyrimidine derivatives from chalcones. *Pharm Chem* 2014, 6, 352.
- Zhang, L., Peng, X., Damu, G. L. V, Geng, R. & Zhou, C. Comprehensive review in current developments of imidazole-based medicinal chemistry. *Med. Res. Rev.* 2014, 34, 340–437.
- Vishal, D. J., Mahendra, D. K. & Sarita, S. Synthesis and pharmacological study of some novel pyrimidines. *Pelagia Res. Libr.* 2012, 3, 343–348.
- Bano, T., Kumar, N. & Dudhe, R. Free radical scavenging properties of pyrimidine derivatives. *Org. Med. Chem. Lett.* 2012, 2, 34.
- Rahman, M. A. Chalcone: A valuable insight into the recent advances and potential pharmacological activities. *Chem Sci J.* 2011, 29, 1–16.
- Gaonkar, S. L. & Vignesh, U. N. Synthesis and pharmacological properties of chalcones: a review. *Res. Chem. Intermed.* 2017, 43, 6043–6077.
- Bhat, B. A. et al. Synthesis and biological evaluation of chalcones and their derived pyrazoles as potential cytotoxic agents. *Bioorg. Med. Chem. Lett.* 2005, 15, 3177–3180.
- Nowakowska, Z. A review of anti-infective and anti-inflammatory chalcones. *Eur. J. Med. Chem.* 2007, 42, 125–137.
- Xue, C. X., Cui, S. Y., Liu, M. C., Hu, Z. D. & Fan, B. T. 3D QSAR studies on antimalarial alkoxyated and hydroxylated chalcones by CoMFA and CoMSIA. *Eur. J. Med. Chem.* 2004, 39, 745–753.
- Lavanya, D. *JOURNAL OF INTERNATIONAL PHARMACE. J. Int. Pharm. Sci.* 2016, 3, 1.
- Sahu, K. K. et al. Applying ultra-accelerated quantum chemical molecular dynamics technique for the evaluation of ligand protein interactions. *Med. Chem. Res.* 2010, 19, 1–10.
- Zhang, X.-W. et al. Synthesis and evaluation of antiinflammatory activity of substituted chalcone derivatives. *Med. Chem. Res.* 2010, 19, 403–412.
- Yadav, H. L., Gupta, P., Pawar, R. S., Singour, P. K. & Patil, U. K. Synthesis and biological evaluation of anti-inflammatory activity of 1, 3 diphenyl propenone derivatives. *Med. Chem. Res.* 2011, 20, 461–465.
- Vogel, S., Ohmayer, S., Brunner, G. & Heilmann, J. Natural and non-natural prenylated chalcones: synthesis, cytotoxicity and anti-oxidative activity. *Bioorg. Med. Chem.* 2008, 16, 4286–4293.
- Yayli, N. et al. Synthesis and biological activities of N-alkyl derivatives of o-, m-, and p-nitro (E)-4-azachalcones and stereoselective photochemistry in solution, with theoretical calculations. *Turkish J. Chem.* 2006, 30, 505–514.
- K Sahu, N., S Balbhadra, S., Choudhary, J. & V Kohli, D. Exploring pharmacological significance of chalcone scaffold: a review. *Curr. Med. Chem.* 2012, 19, 209–225.
- Vasil'ev, R. F., Kancheva, V. D., Fedorova, G. F., Batovska, D. I. & Trofimov, A. V. Antioxidant activity of chalcones: The chemiluminescence determination of the reactivity and the quantum chemical calculation of the energies and structures of reagents and intermediates. *Kinet. Catal.* 2010, 51, 507–515.
- Bag, S., Ramar, S. & Degani, M. S. Synthesis and biological evaluation of α , β -unsaturated ketone as potential antifungal agents. *Med. Chem. Res.* 2009, 18, 309–316.
- Lahtchev, K. L., Batovska, D. I., St P, P., Ubiyovk, V. M. & Sibirny, A. A. Antifungal activity of chalcones: a mechanistic study using various yeast strains. *Eur. J. Med. Chem.* 2008, 43, 2220–2228.
- MT Albuquerque, H., MM Santos, C., AS Cavaleiro, J. & MS Silva, A. Chalcones as Versatile Synthons for the Synthesis of 5-and 6-membered Nitrogen Heterocycles. *Curr. Org. Chem.* 2014, 18, 2750–2775.
- Kaushik, S., Kumar, N. & Drabu, S. Synthesis and anticonvulsant activities of phenoxychalcones. *Pharma Res.* 2010, 3, 257–262.
- Amanullah, M., et al. Cytotoxic, antibacterial activity and physico-chemical properties of some acid catalyzed Schiff bases. *African J. Biotechnol.* 2011, 10, 209–213.
- Hakimi, M., Kukovec, B. M. & Minoura, M. 2D s-d Mixed-Metal Coordination Polymer Containing Potassium, Chromium (III) and Dipicolinate Ions: Preparation and Crystal Structure. *J. Chem. Crystallogr.* 2012, 42, 290–294..