

Synthesis and Characterization of some New synthesis of N-(pyrimidin-2-yl)benzenesulfonamide derivatives combined with oxaimidizolidine

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Abstract

This work involved prepared of some new series of chemical compounds of 1,3-oxazepine and 1,3-Diazepine derivatives in scheme [I]. First step reaction of sulfadiazine with 4-amino acetophenone product of Schiff base (A) 4-((1-(4-aminophenyl)ethylidene)amino)-N-(pyrimidin-2-yl)benzene sulfonamide. Then, Schiff base [A] enter reactions one of with glycine to product five-member ring (B) 4-(2-(4-aminophenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide. compounds (B) were condensed with different aromatic aldehydes such as (p-hydroxybenzaldehyde, 4-methoxybenzaldehyde and salicylaldehyde) in ethanol absolute absolute to Produce new benzanils derivatives [B₁-B₃] respectively, Then the Produce of benzanils derivatives [B₁-B₃] were enter reaction with phthalic anhydride in toluene to Produce new N-(pyrimidin-2-yl)benzenesulfonamide derivatives [B_{1a}-B_{3a}] ,reaction compound [B_{2a}] with Naphthyl amine to give 2-hydroxyphenyl)-4-(naphthalen-2-yl)-1,5-dioxo-3,4,5,5a-tetrahydro-1H-benzo[e][1,3]diazepin-2(9aH)-yl)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl) benzenesulfonamide [C]. All compounds were characterized by FT-IR spectroscopy and M.P, some of them were characterized by ¹H-NMR and spectroscopy analysis.

Keyword :- Sulfadiazine, 4-aminoacetophenone, oxaimidizolidine, 1,3-Oxazepine, 1,3-Diazepine

INTRODUCTION:-

Sulfadiazine is N-substituted derivative of the parent compound, sulfanilamide . antibacterial drug used typical sulfonamide structure ⁽¹⁾, and Antihypertensive drugs agents can be design by conjunction sulfadiazine and Antihypertensive drugs in one compound ⁽²⁾. In 1864 the German chemist Hugo Schiff described the formation of N-substituted imines so they are called (Schiff base) . the synthesis via density of primary amines with effective carbonyls Hydrazides derivatives have growing importance because of the wide spectrum of their biological applications like antibacterial, antitumoral, anti-inflammatory, antifungal and antitubercular agents⁽³⁾ . Oxazepine is Heterogeneous seven member ring that contains two heteroatom (Oxygen and Nitrogen). Diazepine is an analogue to oxazepine and thiazepine but the difference is nitrogen, oxygen, Sulphur atom, Diazepam (valium) is a substituted benzodiazepine introduced in 1964 which was used for the organization of tension and anxiety states, the extenuation of muscle convulsion ⁽⁴⁾ . in medical Heterocyclic compounds has important, agrarian and manufacturing from these heterocyclic that contains nitrogen atom in it's structure , which specialized in good properties as a drugs and repeller , polymers and dyes. There for some important pharmaceuticals which contains oxazepine used as a convulsant drug with bleakness and schezofrenic diseases and anti convulsive, antithrob⁽⁵⁻⁶⁾ . This research involved preparation some new heterocyclic derivative like oxazole, oxazepine, diazapine.

SYNTHESIS METHODS

Procedure: Preparation of N-(pyrimidin-2-yl)benzene sulfonamide (A, B₁, B₂ & B₃) ⁽⁷⁾.

- 1- Dissolve of (2.5 mg) from sulfadiazine with (1.35mg) of 4-amino acetophenone in 40mL of ethanol absolute .
- 2- Add to the previous solution two drops from glacial acetic acid .
- 3- The mixture refluxed for (20) hour at a temperature 78C⁰.
- 4- collect the product by filtration .
- 5- the reaction showed by TLC that completed by using (ethyl acetate:toluene, 1 :4).

1- 4-((1-(4-aminophenyl)ethylidene) amino)-N-(pyrimidin-2-yl)benzenesulfonamide[A].

2- 4-(2-(4-((4-hydroxybenzylidene)amino)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide[B₁].

3- 4-(2-(4-((2-hydroxybenzylidene)amino)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide[B₂].

4- 4-(2-(4-((4-methoxybenzylidene)amino)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide[B₃].

Procedure:- of 4-(2-(4-aminophenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzene sulfonamide[B] ⁽⁸⁾.

To the mixture of Schiff bases [A] (0.01 mol) in THF (20 mL), was added glycine (0.01 mol, 1.48 mg) . mixture was stirred for (6 hour) when temperature (50 °C). The separated solid was dried and re-crystallized by mixture of (75% THF + 25% EtOH).

Procedure:-for perparation of 1,3-oxazepine-4,7-dione derivatives [B_{1a}- B_{3a}] ⁽⁹⁻¹⁰⁾.

A mixture of equal mole from compound [B₁- B₃] and appropriates phthalic anhydride in (250 mL) of toluene, The solution was stirred for (15 hour) . Heating at Boiling point of the solvent, The residue were collected by filtration and the product colored crystalline solid was Crystal restoration from dry 1,4-dioxan . the reaction showed by TLC that completed by using (ethyl acetate : toluene , 1: 4).

1-synthesis of 4-(2-(4-(3-(4-hydroxy-2-methylphenyl)-1,5-dioxo-1,5,5a,9a-tetrahydrobenzo[e][1,3]oxazepin-4(3H)-yl)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide[B_{1a}].

2-synthesis.of.4-(2-(4-(3-(2-hydroxyphenyl)-1,5-dioxo-1,5,5a,9a-tetrahydrobenzo[e][1,3]oxazepin-4(3H)-yl)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide [B_{2a}].

3-Synthesis.of.4-(2-(4-(3-(4-methoxyphenyl)-1,5-dioxo-1,5,5a,9a-tetrahydrobenzo[e][1,3]oxazepin-4(3H)-yl)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide[B_{3a}].

Procedure:-for perparad of 1,3-diazepine-4,7-dione[C] ⁽¹¹⁾.

A mixing up equal mole from compounds oxazepine and naphthyl amine in toluene(30mL) in round bottom flask. The solution was stirred for (5 hour) . Heating at Boiling point of the solvent then allowed to cool to room temperature and separated precipitate was filtered and Crystal restoration from ethanol. the reaction showed by TLC that completed by using (ethyl acetate : toluene , 1 : 4).

1- 4-(2-(4-(3-(2-hydroxyphenyl)-4-(naphthalen-2-yl)-1,5-dioxo-3,4,5,5a-tetrahydro-1H-benzo[e][1,3]diazepin-2(9aH)-yl)phenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzenesulfonamide[C].

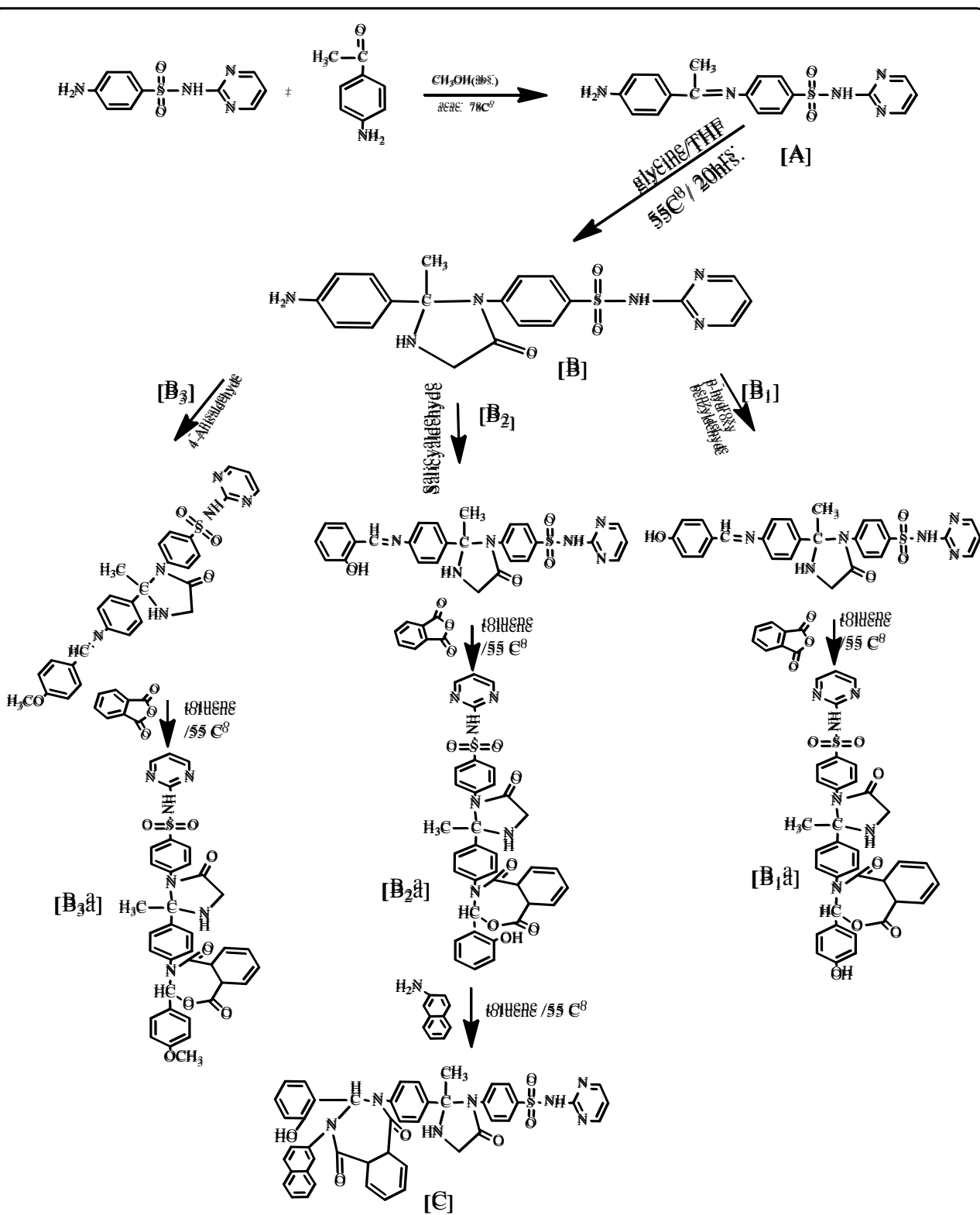


Table (1) some physical properties of compounds [A-C].

Comp. No.	Structural formula	Molecular Formula	M.P °C	Yield %	R _f
A		C ₁₈ H ₁₇ N ₅ O ₂ S	276-278	90	0.65
B		C ₂₀ H ₂₀ N ₆ O ₃ S	259-261	85	0.56
B ₁		C ₂₇ H ₂₄ N ₆ O ₄ S	276-277	86	0.58
B ₂		C ₂₇ H ₂₄ N ₆ O ₄ S	269-271	80	0.72
B ₃		C ₂₈ H ₂₆ N ₆ O ₄ S	265-266	82	0.68
B _{1a}		C ₃₆ H ₃₄ N ₆ O ₇ S	280-281	80	0.63
B _{2a}		C ₃₅ H ₃₀ N ₆ O ₇ S	300-301	84	0.75
B _{3a}		C ₃₆ H ₃₂ N ₆ O ₇ S	264-265	87	0.68
C		C ₄₅ H ₃₇ N ₇ O ₆ S	250-251	81	0.72

Table [2] FT-IR data of Schiff bases compounds [A-B₃].

Comp.No.	Ar	ν (C=N) Imine cm ⁻¹
A		1576
B ₁		1575
B ₂		1578
B ₃		1577

Table [3] FT-IR data of compound [B₁a- B₃a] .

Comp No.	Ar	ν (C=C) Aromatic cm ⁻¹	ν (C-H) Oxazepine ring cm ⁻¹	ν (C=O)str. Lactone Lactam cm ⁻¹	ν (C-N) cm ⁻¹	ν (C-O) Lactone cm ⁻¹
B ₁ a		1410 1442	3109	1653 1704	1160	1282
B ₂ a		1405 1449	3036	1762 1791	1168 1257	1281
B ₃ a		1404 1440	3024	1761 1790	1167	1257

DISCUSSION:-**3.1- Synthesis and Identification of Schiff bases Derivatives [A-B₁,B₂,B₃]:**

Schiff bases were synthesized by the condensation reactions of different aromatic aldehydes such as p-hydroxybenzaldehyde, 3-methoxy benzaldehyde, salicylaldehyde with amine derivatives in the existence of glacial acetic acid as catalyst in absolute ethanol. The preparation compounds [A-B₁,B₂,B₃] were characterized by FT-IR which showed band at (1575-1588) cm⁻¹ of stretching vibration of imine group (C=N) ^(12,13). Other information of functional groups.

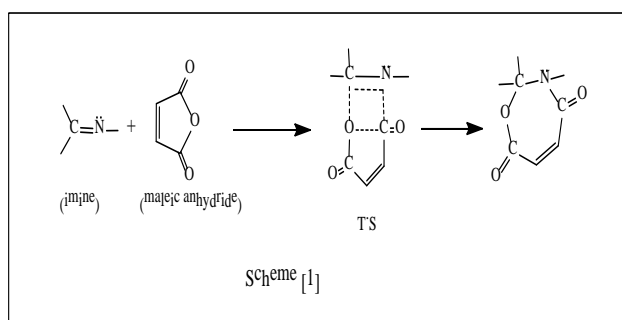
Synthesis and characterization of 4-(2-(4-aminophenyl)-2-methyl-5-oxoimidazolidin-1-yl)-N-(pyrimidin-2-yl)benzene sulfonamide[B].

The compound [B] was synthesis by the reaction of Schiff base [A] with the amino acid glycine in THF .

The FT-IR spectra of the compounds [B] showed disappearance of absorption bands at (1576) cm⁻¹ was return the (C=N) of imine group and appearance of absorption band at (1647 lactam – 1707 lactone) ⁽¹⁴⁾ cm⁻¹ was return the stretching vibration from the group(C=O). in compound (imidazolidin), & the emergence from two packages in signatories (2924 – 3039) cm⁻¹ which dated back to the stretching frequency (C-H) aromatic and aliphatic respectively.

synthesis and Identification of 1,3-oxazepine-4,7-dione derivatives [B₁a- B₃a].

Pericyclic reactions, between imine groups of schiff bases [B₁-B₃] and cyclic acid anhydride [maleic anhydride] in toluene, were carried out to the synthesis of 1,3-oxazepine derivatives [B₁a-B₃a]. Mechanism ⁽¹⁵⁾ of the pericyclic reaction for the preparation seven membrane shown in scheme[1].



The synthesized compounds [B₁a-C₅b] were characterized by FT-IR spectra, some of them were describe by ¹H-NMR spectra .

The FT-IR spectra of the compounds [B₁a-C₃b] showed appearance of the powerful absorption band at (1701-1790) cm⁻¹ was due to (C=O) lactone group the stretching vibration ⁽¹⁶⁾, the appearance of the stretching vibration of the (C=O) lactam group at (1653-1762) cm⁻¹ ⁽¹⁷⁾. Other information of functional groups were shown at following table [2].

Synthesis and Identification of for Synthesis of 1,3-diazepine-4,7-dione [C].

1,3-Diazepine derivatives were prepared from reaction between 1,3-oxazepine with sulphadiazine in dry benzene and the following compounds are prepared [C]. Mechanism ⁽¹⁸⁾ of the synthesis 1,3-diazepine ring is shown in scheme [4].

FT-IR spectra describe The synthesized compounds [C] the stretching vibration of the (C=O) lactone group showed disappearance at (1651-1652) cm⁻¹ ⁽¹⁹⁾, the stretching vibration of the (C=O) lactam group the appearance of the strong absorption band at (1699-1710) cm⁻¹ ⁽²⁰⁾, following table [3].

¹H-NMR spectrum of compounds [B₁ and C] showed the following characteristic signals (DMSO-*d*₆ as a solvent) the multiplet signal at δ (7.6-8.1 ppm) that could be attributed to the aromatic protons for ten phenyl rings and the doublet signal at δ (6.5-7.1)ppm that could be attributed to the two protons of seven membered ring of oxazepine(2H of double bond of oxazepine ring) group The ¹H-NMR spectrum also showed the singlet signal at δ (7.3-7.9 ppm) that could be attributed to the one proton of oxazepine(CH of oxazepine ring) group ⁽²¹⁾ and other data of groups containing protons were showed in table [4].

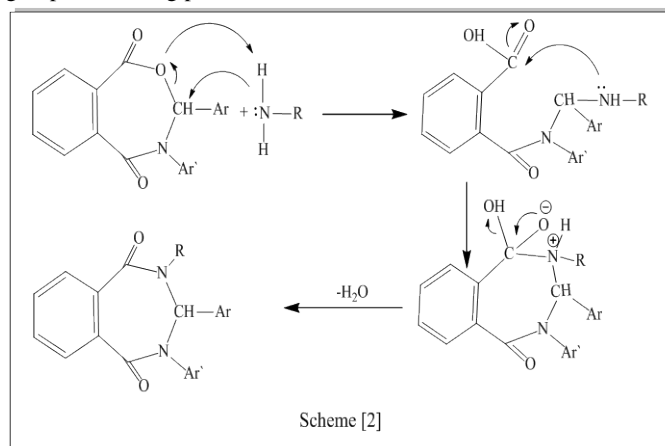
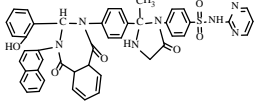
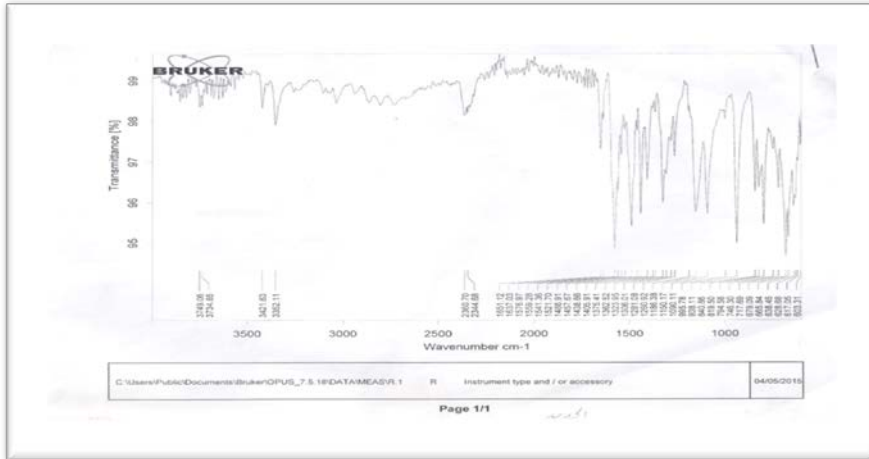
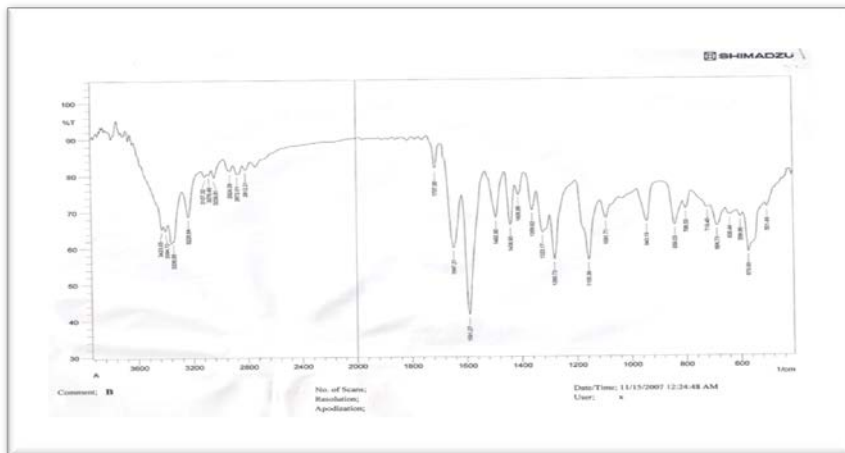


Table [4] FT-IR data of compound [C].

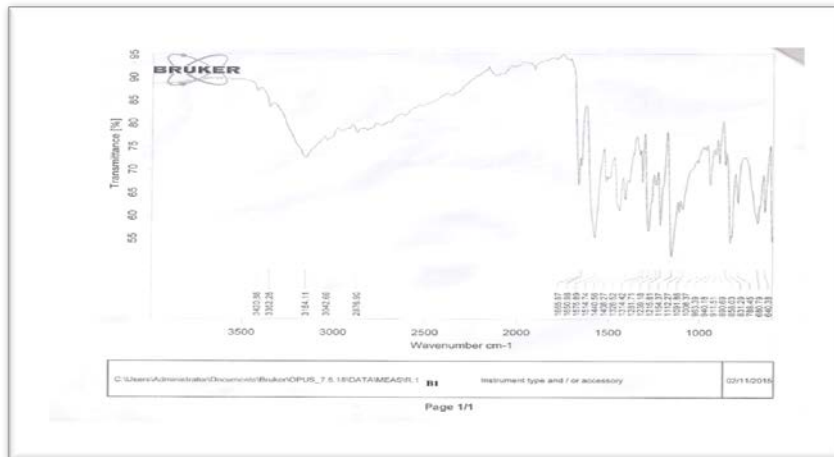
Comp. No.	Ar	ν (C-H) oxazepine ring cm^{-1}	ν (C=O)str. Lactone Lactam cm^{-1}	ν (C-N) Lactone cm^{-1}	ν (NH) bending cm^{-1}	Others cm^{-1}
C		3225	1652 1699	1288	1458 1485	ν (C-OH) :3340



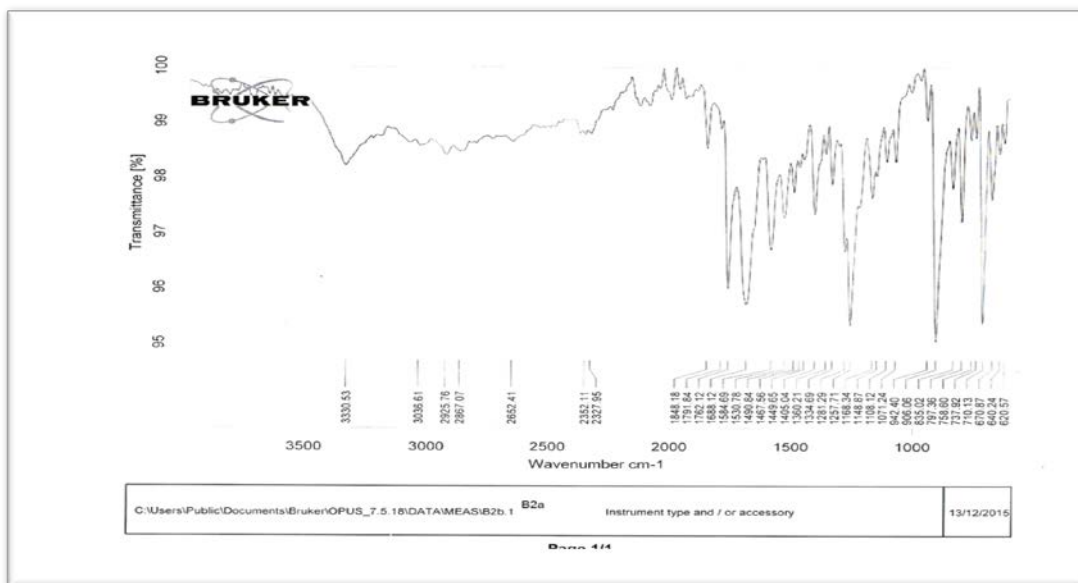
FT-IR spectrum of compound [A]



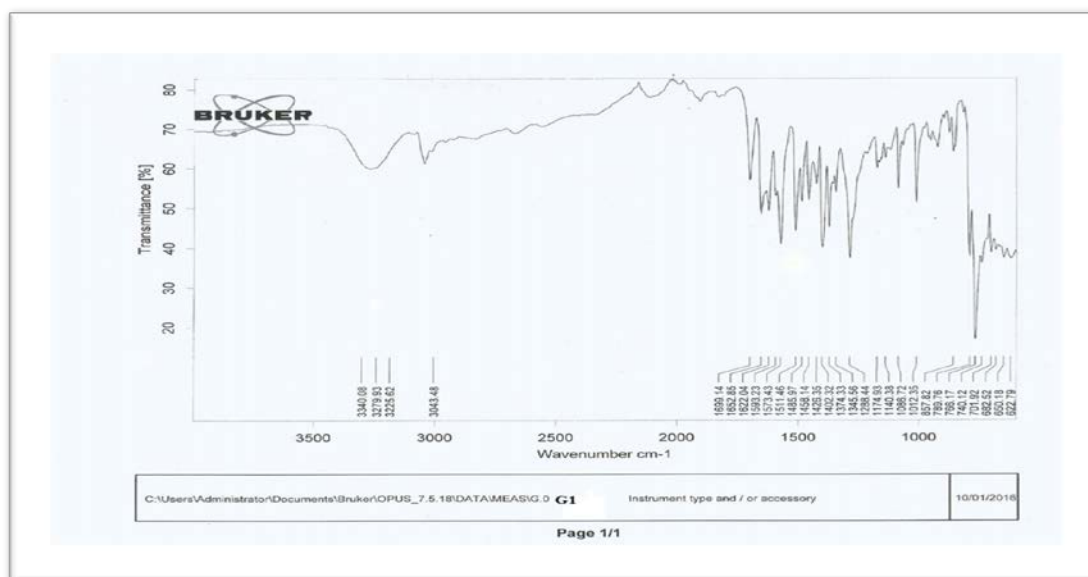
FT-IR spectrum of compound [B]



FT-IR spectrum of compound [B₁]

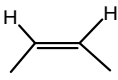
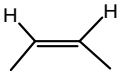


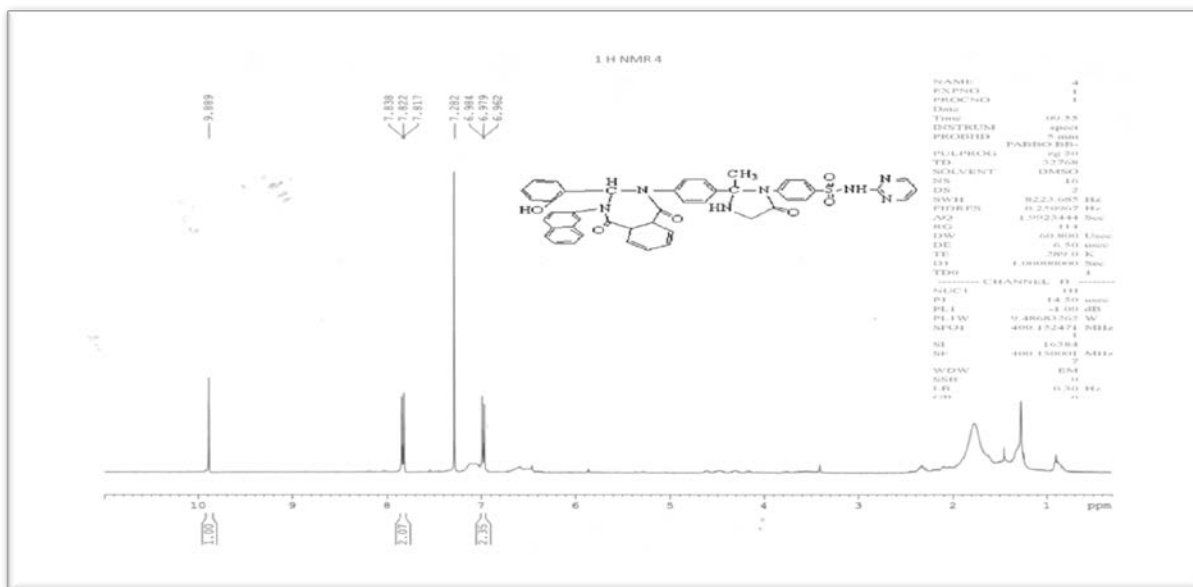
FT-IR spectrum of compound [B₂a]



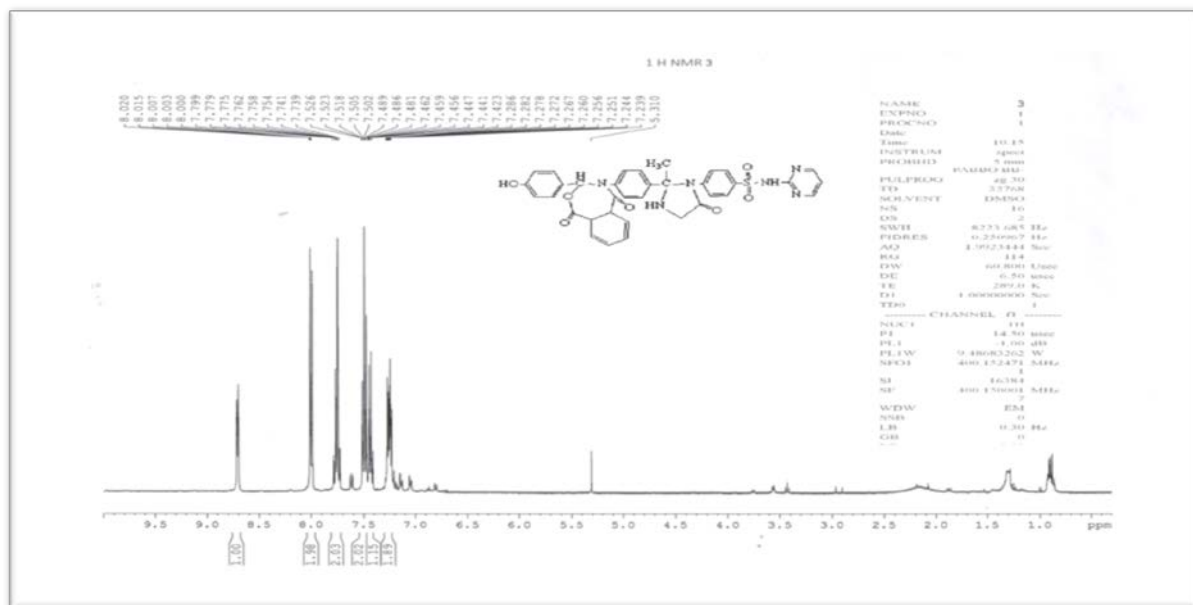
FT-IR spectrum of compound [C]

Table [4] H-NMR data of compound [B₁a-C].

Comp. No.	$\delta(\text{C-H})$ Aromatic ppm		CH of oxazepine ring	NH Sulphadiazene	HC=N Pyrimidine	$\delta(\text{C-H})$ of CH ₃ <i>o</i> -tolidine	Others Ppm
B ₁ a	7.6	6.5-6.7	7.3	11.3	8.5	2.5	(s, 6H, N-CH ₃): δ 1.8
Comp. No.	$\delta(\text{C-H})$ Aromatic		CH of Diazepine ring	3H of CH ₃ <i>o</i> -tolidine	Others Ppm		
C	8.1	6.9-7.1	7.9	2.5	–		



H-NMR spectrum of compound [C]



H- NMR spectrum of compound [B]a

CONCLUSIONS:

The study arrived at the following Conclusions:

- 1-The electron-donating and the electron-withdrawing groups affect the determination of the time of the reaction. The electron-donating group increases the rate of the reaction, therefore the time of the reaction decreases. While the electron-withdrawing group decreases, the rate of the reaction, therefore, the time of the reaction was increases'.
- 2-All synthesized compounds were stable by resonance and having high melting points relatively; this is another evidence in relation to stability.
- 3-Pericyclic reactions, between imine groups and maleic anhydride, phthalic anhydride were carried out to the synthesis of 1,3-oxazepine derivatives.

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