

New Derivatives of Thiozolidinone, Synthesis and Characterization

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Abstract

The present work involved synthesis of new thiozolidinone derivatives. These derivatives could be divided into three types of compounds; quinolin-2-one [V]_{a,b}, Schiff bases [VI]_{a,b} and imide compounds [VII]_{a-d}. The reaction of p-Hydroxyacetophenone with thiosemicarbazide led to the formation of thiosemicarbazone compound [II], which reacted with chloroacetic acid in CH₃CO₂Na to yield 4-thiazolidinone compound [III]. In addition, thiosemicarbazide was reacted with POCl₃ to give [IV] compound used as intermediates to synthesize new compounds of reacted with two types of coumarin in glacial acetic acid to give quinolin-2-one [V]_{a,b}. The later compound refluxed with different benzaldehyde in dry benzene and glacial acetic acid to give Schiff bases [VI]_{a,b}. While the reaction with four kinds of acid anhydride using dioxane afforded imide compounds [VII]_{a-d}. The synthesized compounds were identified using FTIR and ¹H NMR spectroscopy.

Key Words: thiozolidinone, thiadiazol, quinolin-2-one, Schiff bases and imide compounds.

INTRODUCTION

4-Oxothiazoles have a wide range of biological activities depending on their structure. Lately, many workers interested in the synthesis of new 4-oxothiazolidine-2,5-diyldene derivatives containing significant pharmacophore groups, such as thiazole, benzothiazole, benzimidazole⁽¹⁾. Thus, compounds have high anti-malarial⁽²⁾ and anti-tumor activity⁽³⁾. Among the heterocyclic compounds, 1,3,4-thiadiazole derivatives have become an important type for the development of new drugs⁽⁴⁾. The compounds which contain 1,3,4-thiadiazole appeared a good biological activity spectrum⁽⁵⁾ including anti-microbial⁽⁶⁾, anti-tuberculosis⁽⁷⁾, anti-inflammatory⁽⁸⁾, carbonic anhydrase inhibitors⁽⁹⁾, anti-convulsants⁽¹⁰⁾, anti-hypertensive⁽¹¹⁾, anti-oxidant⁽¹²⁾, anti-cancer⁽¹³⁾ and anti-fungal properties⁽¹⁴⁾. 1,3,4-thiadiazole heterocyclic compounds undergo various chemical reactions which have made them important for new molecule planning because of their lucky structure, which has many biological potentials⁽¹⁵⁾. As an example, two 1,3,4-thiadiazole compounds currently used in clinical medicine are: acetazolamide and methazolamide as carbonic anhydrase inhibitors⁽¹⁶⁾.

The number of scientific studies with imide compounds has increased considerably⁽¹⁷⁾. Taking into account the importance of imide compounds in both heterocyclic chemistry and medicinal fields⁽¹⁸⁾. In recent years the study of the chemistry of imides has been given specific importance, because of their pharmacological and other industrial uses. For examples, derivatives of imides have been evidenced to be important agents and have been used in the treatment of arthritis, tuberculosis, convulsions and epilepsy⁽¹⁹⁾. Some derivatives of imides can be used to stimulate the growth of plants⁽²⁰⁾. Other imide derivatives have been used as fungicides and as herbicides⁽²¹⁾. While the aromatic imides are used as brightening agents in the laundry and allied industries^(22,23). Pyrrolidine-2,5-diones are an important class of heterocyclic compounds with good applications in organic synthesis and medicinal chemistry^(24,25).

Among the pharmacologically important heterocyclic compounds, quinoline and its derivatives have been

known to possess various biological activities,^(26,27) and quinolines are active components in various industrial antioxidants and dyes.⁽²⁸⁻³⁰⁾

The aim of this work is the synthesis and characterization of new thiazolidinone compounds and their derivatives (Schiff bases, quinoline and imide compounds).

MATERIALS AND METHODS

Preparation of 2-(1-(4-hydroxyphenyl)ethylidene)hydrazine-1-carbothioamide [II]

A mixture of 4-hydroxyacetophenone [I] (1.36g, 0.01 mole), thiosemicarbazide (0.91g, 0.01 mole) in ethanol (20 mL) was refluxed for 5 hrs⁽³¹⁾, after cooling, the solid was filtered, dried and recrystallized from ethyl acetate.

Color: White; Yield: 78%; M.P: 219-221.⁽³²⁾

IR (ν, cm⁻¹): 3365-3178 (NH₂NH), 3014 (CH arom.), 2918-2850 (CH aliph.), 1645 (C=N), 1600 (C=C), 1282 (C=S).

Synthesis of 2-(4-(1-((4-oxothiazolidin-2-ylidene)hydrazono)ethyl)phenoxy)acetic acid [III]

Heated a mixture of thiosemicarbazone [II] (2.09g, 0.01 mol), chloroacetic acid (1.88g, 0.02 mol) and sodium acetate (fused) (4.92g, 0.06 mol) in absolute ethanol (10 mL) for 6 hrs⁽³³⁾. Then, the mixture of reaction was poured onto (100 mL) cold water and the precipitate was filtered, washed with water (for many times), recrystallized from ethanol.

Color: Off white; Yield: 60%; M.P: 240-242^oC.

IR (ν, cm⁻¹): 3340-2600 (OH), 3234 (NH), 3050 (C-H arom.), 2972-2933 (C-H aliph.), 1691 (C=O), 1629 (C=N), 1600 (C=C), 796 (C-S).

¹H NMR (δ ppm): 2.32 (s, 5H, CH₂ cyclic and CH₃), 3.8 (s, 2H, OCH₂), 6.78-7.72 (dd, 4H, Ar-H), 9.85 (s, 1H, NH), 11.85 (s, 1H, OH).

Synthesis of 2-((1-(4-((5-amino-1,3,4-thiadiazol-2-yl)methoxy)phenyl)ethylidene)hydrazono)thiazolidin-4-one [IV]

Compound [III] (3.07g, 0.01 mol) was mixed with thiosemicarbazide (0.91g, 0.01 mol) in phosphorus oxy

chloride (5 mL), afterward the mixture was refluxed softly for 6 hrs⁽³⁴⁾. After cooling, was poured onto ice water (50 mL) with stirring. The precipitate was filtered, washed with solution of NaHCO₃, then water, dried and recrystallized from ethanol.

Color: Brown; Yield: 80%; M.P:> 300⁰C

IR (v, cm⁻¹) :3360-3232(NH₂), 3020(C-Haromatic), 2924-2840 (C-H aliphatic), 1697 (C=O), 1660(C=N), 1604 (C=C), 777(C-S) .

¹H NMR (□ppm): 2.36 (s, 5H,CH₂ cyclic and CH₃), 3.84 (s, 2H,OCH₂), 5.22(broad s, 2H,NH₂), 6.8-7.84 (m, 4H, Ar-H),11.7(s, 1H,NH).

SYNTHESIS OF QUNOLINE DERIVATIVES

Equimolar amounts of two types from coumarin (0.01 mol) and amine compounds [IV] (3.62g,0.01mol) in glacial acetic acid (10mL) was heated under reflux for 6hrs. After cooling poured onto crushed ice to afford, the resulting was filtered and dried at room temperature. Recrystallization from ethanol.

2-((1-(4-((5-(2-oxoquinolin-1(2H)-yl)-1,3,4-thiadiazol-2-yl)methoxy)phenyl) ethylidene)hydrazono)thiazolidin-4-one[V]_a

Color: Gray; Yield: 61%; M.P: 238-240⁰C

IR (v, cm⁻¹) :3059(C-Haromatic),2953-2850(C-H aliphatic), 1701,1670 (C=O), 1616(C=N), 1600(C=C),725(C-S).

¹H NMR (□ppm):2.31 (s, 5H,CH₂ cyclic and CH₃), 3.85 (s, 2H,OCH₂), 6.48(d,2H,CH of CH=CH),6.78 (m, 8H, Ar-H),11.79(s, 1H,NH) .

2-((1-(4-((5-(7-hydroxy-4-methyl-2-oxoquinolin-1(2H)-yl)-1,3,4-thiadiazol-2-yl)methoxy)phenyl)ethylidene)hydrazono)thiazolidin-4-one[V]_b

Color:Brown; Yield: 83%; M.P: 190-192⁰C

IR (v, cm⁻¹) :3300(OH), 3100(C-Haromatic),2950-2819(C-H aliphatic), 1700,1666 (C=O), 1620(C=N), 1598(C=C),748(C-S).

¹H NMR (□ppm): 2.2(s,3H,CH₃at C4 of ring), 2.37 (s, 5H,CH₂ cyclic and CH₃C=N), 3.85 (s, 2H,OCH₂), 6.13(s,1H,OH), 6.41(s,1H,CH=C) 6.65-8.0 (m, 7H, Ar-H),10.56(s, 1H,NH) .

SYNTHESIS OF SCHIFF BASE DERIVATIVES

The mixture of amino compound[IV] (0.01 mol) and different aldehyde (0.01 mol) in benzene (5 mL) addition 3 drops of glacial acetic acid was refluxed for 6h. The solvent was evaporated under vacuum and the residue crystallized from ethanol.

2-((1-(4-((5-(benzylideneamino)-1,3,4-thiadiazol-2-yl)methoxy)phenyl) ethylidene)hydrazono)thiazolidin-4-one[VI]_a

Color:Brown; Yield: 92%; M.P: 153-155⁰C

IR (v, cm⁻¹) :3100(C-Haromatic),2926-2852(C-H aliphatic), 1708 (C=O), 1681(C=N), 1598(C=C) ,754(C-S).

¹H NMR (□ppm): 2.35 (s, 3H,CH₃),2.48(s,2H, CH₂ cyclic), 3.9(s, 2H,OCH₂), 6.87-7.94 (m, 9H, Ar-H),8.4(s,1H,CH imine)11.43(s, 1H,NH).

2-((1-(4-((5-((4-methoxybenzylidene) amino)-1,3,4-thiadiazol-2-yl)methoxy)

phenyl)ethylidene)hydrazono)thiazolidin-4-one[VI]_b

Color:Brown; Yield: 96%; M.P: 230-232⁰C

IR (v, cm⁻¹) :3120(C-Haromatic),2929-2839(C-H aliphatic), 1693 (C=O), 1678(C=N), 1595(C=C) ,1247(C-O-C),759(C-S).

¹H NMR (□ppm):2.3 (s, 3H,CH₃),2.48(s,2H, CH₂ cyclic), 3.65(s, 2H,OCH₂), 3.88(s, 3H,OCH₃),6.87-7.89 (m, 8H, Ar-H),8.34(s,1H,CH imine)9.88(s, 1H,NH).

SYNTHESIS OF IMIDE DERIVATIVES [VII]_{A-D}

Refluxed a mixture of amine compound [III] (0.001mole) and different anhydride (0.002mole) in dry dioxane (20 ml) for 5 hrs, the reaction mixture was left overnight for slow evaporation. The product was recrystallized from acetone.

1-(5-((4-(1-((4-oxothiazolidin-2-ylidene)hydrazono)ethyl)phenoxy)methyl)-1,3,4-thiadiazol-2-yl)pyrrolidine-2,5-dione[VII]_a

Color:Gray; Yield: 83%; M.P: 170-172⁰C

IR (v, cm⁻¹) :3080 (C-H) aromatic, 2927-2820(C-H aliphatic),1708,1699,1680(C=O),1625(C=N),1598 (C=C) aromatic, 1359(C-N) .

¹H NMR (□ppm): 2.31(t,4H,CH₂-CH₂ cyclic), 2.43(s, 5H,CH₂ cyclic and CH₃), 3.7 (s, 2H,OCH₂), 6.80-7.8 (m, 4H, Ar-H),12(s, 1H,NH).

2-(5-((4-(1-((4-oxothiazolidin-2-ylidene)hydrazono)ethyl)phenoxy)methyl)-1,3,4-thiadiazol-2-yl)-3a,4,7,7a-tetrahydro-1H-isoindole-1,3(2H)-dione[VII]_b

Color:Brown; Yield: 88%; M.P: 185-187⁰C

IR (v, cm⁻¹) :3050 (C-H) aromatic, 2927-2820(C-H aliphatic),1700,1699,1680(C=O),1628(C=N),1597(C=C) aromatic, 1355(C-N) .

¹H NMR (□ppm): 2.09(d,4H,CH₂-CH₂ cyclic), 2.46(s, 5H,CH₂ cyclic and CH₃), 2.8(d,4H,2CH₂ cyclic), 3.92(s, 2H,OCH₂),5.62(t,2H, CH-CH cyclic),6.75-8.06 (m, 4H, Ar-H),11.8(s, 1H,NH).

2-(5-((4-(1-((4-oxothiazolidin-2-ylidene)hydrazono)ethyl)phenoxy)methyl)-1,3,4-thiadiazol-2-yl)isoindoline-1,3-dione[VII]_c

Color:Brown; Yield: 100%; M.P: 160-162⁰C

IR (v, cm⁻¹) :3100 (C-H) aromatic, 2953-2852(C-H aliphatic),1708,1695,1676(C=O),1620(C=N),1595(C=C) aromatic, 1359(C-N) .

¹H NMR (□ppm):2.31(s, 5H,CH₂ cyclic and CH₃), 4.14 (s, 2H,OCH₂), 6.86-8.05 (m, 8H, Ar-H),12.1(s, 1H,NH).

2-(5-((4-(1-((4-oxothiazolidin-2-ylidene)hydrazono)ethyl)phenoxy)methyl)-1,3,4-thiadiazol-2-yl)-1H-benzof[*f*]isoindole-1,3(2H)-dione[VII]_d

Color:Brown; Yield: 100%; M.P: 240-242⁰C

IR (v, cm⁻¹) :3068 (C-H) aromatic, 2924-2852(C-H aliphatic),1732,1704,1666(C=O),1630(C=N),1600(C=C) aromatic, 1355(C-N) .

¹H NMR (□ppm):2.43(s, 5H,CH₂ cyclic and CH₃), 3.98 (s, 2H,OCH₂), 6.85-8.52 (m, 10H, Ar-H),10.3(s, 1H,NH).

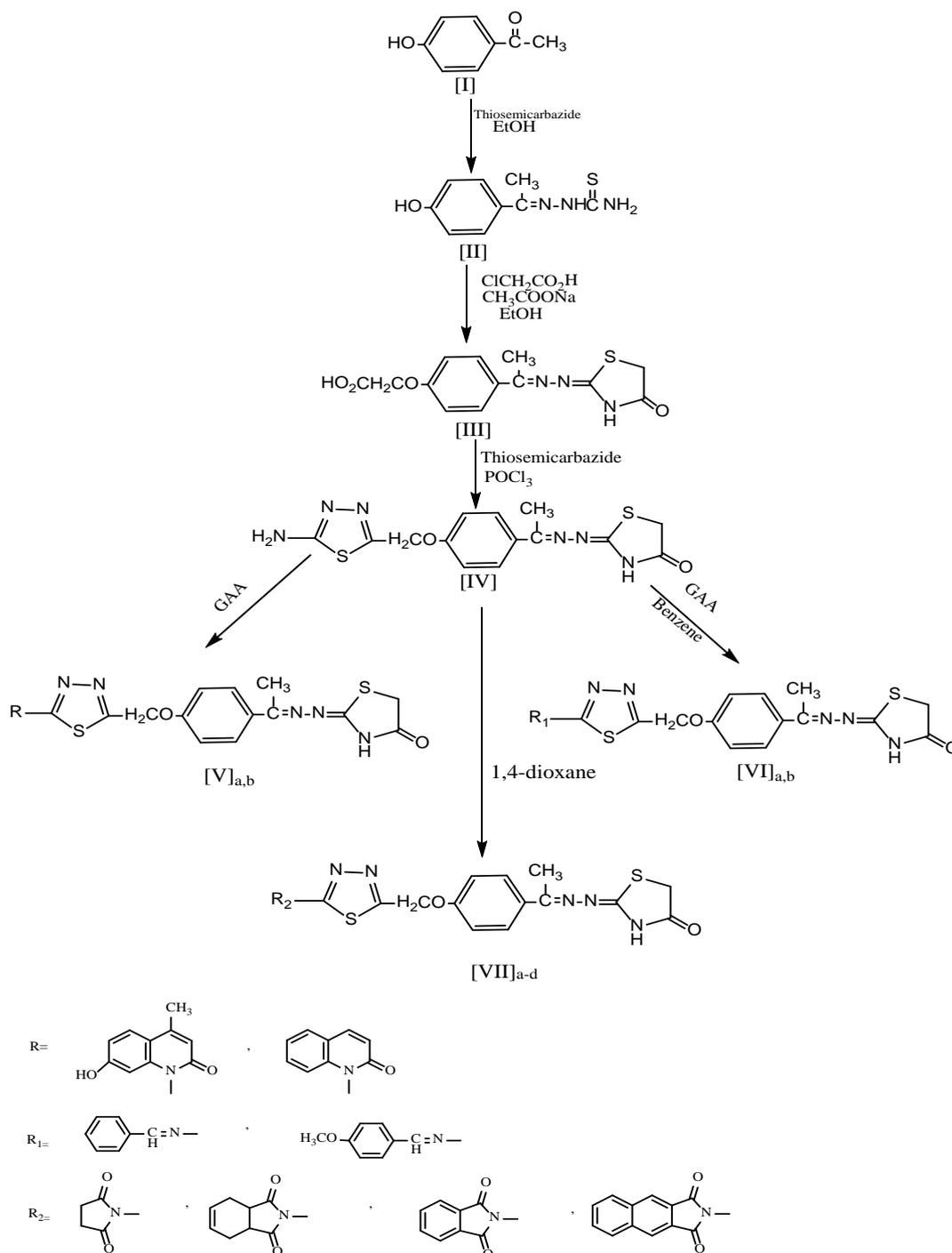


Figure 1: Synthetic Scheme

Analytical Characterization

FTIR spectra (using KBr disc) were recorded by on a Shimadzo (Ir prestige -21), ¹HNMR spectra were recorded by company : Bruker , model: ultra shield 300 MHz , origin :Switzerland (in DMSO as a solvent), ppm(δ), uses internal standard (TMS), were made at chemistry department , Gazi University, Turkey. Hot-Stage, Gallen Kamp melting point apparatus was used for determined uncorrected melting points.

RESULTS AND DISCUSSION

The compound [II] was synthesized from condensation of 4-hydroxyacetophenone with thiosemicarbazide. The compounds [II] was characterized using FTIR spectroscopy and melting point. The FTIR spectrum of compound[II]showed disappearance of ketone ν C=O and appearance ν C=N stretching band. While the compound [III] was synthesized from the refluxing compound [II] with chloroacetic acid in presence of sodium acetate. The FTIR spectrum showed the appearance of new stretching band due to a carbonyl group of thiazolidinone ⁽³⁵⁾and

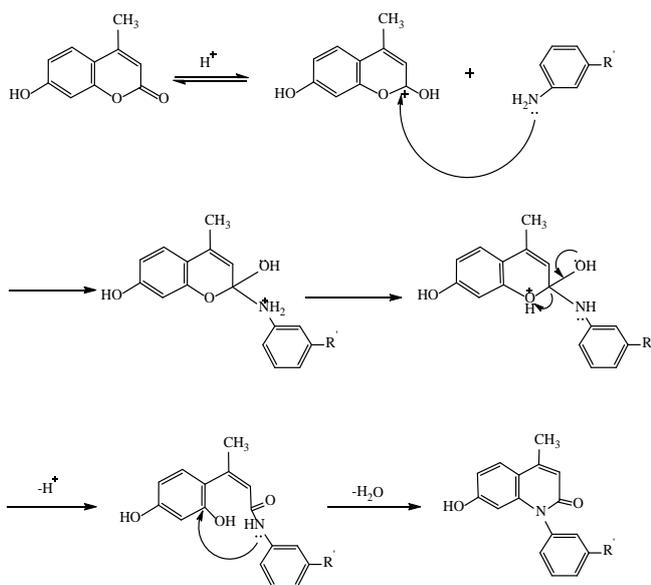
stretching band due to C-S bond, Also appearance a broad stretching band for OH group.

The ^1H NMR spectrum of compound [III] exhibited the following characteristics chemical shifts: two sharp singlet signal could be attributed to protons of NH group and group(OHacid), respectively. Also the spectrum showed two singlet signals for two protons of CH₂ cyclic, and three protons of CH₃ group. Also the spectrum exhibited a singlet signal for two protons of OCH₂ group.

The reaction of thiazolidenone [III] with thiosemicarbazide in POCl₃ yielded a new compound of 1,3,4-thiadiazole [IV]. This compound is identified by FTIR spectroscopy. The FTIR spectrum of compound [IV] showed disappearance absorption band due to OH group and the appearance of a new two bands due to stretching vibration of amine group (NH₂).

While ^1H NMR spectrum showed the following signals: singlet signal for one proton of CO-NH (amide) group. Many signals that could be assigned to the Four aromatic protons, also a singlet signal for two protons thiazolidinone ring and CH₃ group and single to OCH₂ group. Finally a singlet signal appeared for two protons of NH₂ group.

The new quinolin-2-one [V]_{a,b}, were synthesized by refluxing of coumarin with amino compounds [IV]_{a,b} in glacial acetic acid according to the suggested mechanism.



The characteristic FTIR absorption bands of quinolin-2-one [V]_{a,b}, showed a shift in the carbonyl stretching band from lactone group of coumarin to low frequency for lactam group of quinolin-2-one, and disappearance absorption band due to NH₂ group.

The ^1H NMR spectrum of compound [V]a exhibited doublet signal that could be attributed to the CH=CH proton of quinolin-2-one ring, but the ^1H NMR spectrum of compound [V]b showed another two singlet signals for three protons of CH₃ group and a proton of OH group, besides to a singlet signal for CH= proton.

The Schiff bases type [VI]_{a,b} were produced from the refluxing of amino compound [IV] with different aromatic

aldehydes in benzene using three drops of glacial acetic acid (GAA). The FTIR spectra of compounds [VI]_{a,b} showed appearance new absorption band which is assigned to C=N stretching.

The ^1H NMR spectrum of Schiff base [VI]_{a,b} showed eight aromatic protons appeared and a singlet signal for one proton of CH=N group. When treatment amino compound [IV] with different anhydride in dry 1,4-dioxane under reflux led to produce new imide compounds [VII]_{a-d}. Structure of this compound [VII]_{a-d} was confirmed by FTIR, which showed disappearance of absorption bands belong to (-NH₂) amine and C=O of acid anhydride with appearance two stretching bands for C=O of imides.

Finally, the ^1H NMR spectrum of imide compounds [VII]_{a,b} showed singlet signals due to protons of CH₂ or CH groups cyclic moiety for compound [VII]_a and [VII]_b, respectively while the ^1H NMR spectrum of compounds [VII]_{c,d} did not show signals for these groups a proton of CH₂ or CH group cyclic anhydride.

CONCLUSIONS

This work includes synthesis of new derivatives of quinolin-2-one, Schiff bases and imides with good to moderate yields using easy methods.

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